C. ABDUL HAKEEM COLLEGE (AUTONOMOUS), **SEMESTER EXAMINATIONS, NOVEMBER - 2018** MELVISHARAM - 632 509

B.Sc., CHEMISTR\ U15MCH501 - INORGANIC CHEMISTRY - I SEMESTER V

Time: Three Hours Maximum: 75 Marks

SECTION - A $(10 \times 2 = 20 \text{ Marks})$

Answer ALL Questions

- 1. What is DMG? Write its use in gravimetric analysis
- 2. What are the factors that affect the theromogram in TGA?
- 3. What are complex ion and ligands?
- 4. Give IUPAC names to the following a) [Cr (H₂O)₄Cl₂]Cl b) K₃[Fe (CN)₆].
- 5. Define Effective atomic number (EAN) rule. Calculate EAN for [Ni (CO)₄].
- 6. Explain primary valency in Werner's theory
- 7. Write the role of phosphorus in biological activities
- 8. What are metal carbonyls?
- 9. What are pseudo halogens? Give an example
- Arrange the halogens in the order of oxidizing power. Give reason.

SECTION - B (5 X 5 = 25 Marks)

Answer ALL Questions

Ξ. a) Write the principle and important applications of differential thermal

(Or

- b) What are the sources of random errors?
- 12. a) What is chelation? Discuss the applications of EDTA

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b) Explain hydrate isomerism

13. a) Discuss the Splitting of metal d-Orbitals in square complexes

14. a) How Aluminium is estimated using Oxine?

b) Explain low and high spin complexes

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- b) What is Pi-acceptor ligand? Explain compounds of P as acceptor ligands
- 15. a) Account on the similarities and dissimilarities between pseudo halogens and halogens

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b) Compare the general characteristics of halogens.

SECTION - C (3 X10 = 30 Marks)

Answer ANY THREE Questions

- 16. a) Discuss the factors affecting TGA and DTA curves
- b) Explain the instrumentation of the TGA
- 17. a) Explain linkage and polymerization isomerism of Co-ordination Complexes
- Complexes b) Explain why Chelate complexes are more stable than non-Chelate
- 18. a) Explain werner's theory of Co-ordination complexes.
- and account for its diamagnetic character b) Draw the crystal field splitting and energy level diagram of [Ni(CN)₄]²

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- 19. Discuss the application of Co-ordination Compounds in qualitative and quantitative analysis with two examples each.
- 20. a) Give a comparative study about electro negativity, electron affinity and oxidizing power of halogens.
- b) Discuss the important Oxides and Oxy acids of halogens.
